



# Radical [3+2]-cycloaddition reaction with alkenes using dimethyl 2-(iodomethyl)cyclopropane-1,1-dicarboxylate as a new homoallyl radical precursor

Osamu Kitagawa, Hiroki Fujiwara and Takeo Taguchi\*

Tokyo University of Pharmacy and Life Science, 1432-1 Horinouchi, Hachioji, Tokyo 192-0392, Japan

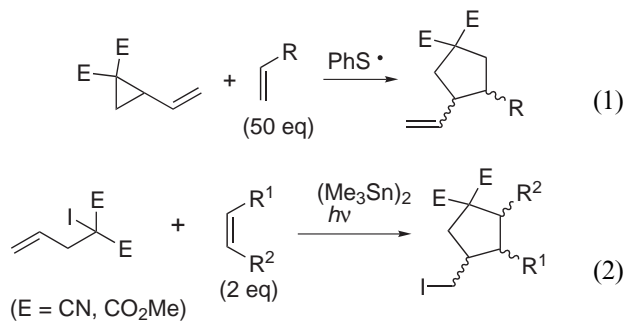
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**Abstract**—Radical iodine atom transfer [3+2]-cycloaddition with various alkenes using dimethyl 2-(iodomethyl)cyclopropane-1,1-dicarboxylate as a new precursor of homoallyl radical species smoothly proceeds to give functionalized cyclopentane derivatives in good yields. © 2001 Elsevier Science Ltd. All rights reserved.

[3+2]-Cycloaddition reaction of a homoallyl radical with an alkene is one of the powerful means for one-step synthesis of a cyclopentanoide skeleton.<sup>1</sup> Radical [3+2]-cycloaddition reaction with simple alkenes (without electron withdrawing groups) has been achieved by the use of electron-deficient homoallyl radicals generated from vinylcyclopropane and allylated  $\alpha$ -iodoactive methine derivatives (Eq. (1) and (2)).<sup>1g-j</sup> However, Feldman and Oshima's method using vinylcyclopropane derivatives as the homoallyl radical precursor required a large excess of simple alkenes (50 equiv.) and could not be applied to the reaction with less reactive 1,2-disubstituted alkenes (Eq. (1)).<sup>1e,j</sup> Although Curran et al. have succeeded in reactions with various 1-alkenes, 1,1- and 1,2-disubstituted alkenes by the use of allyl- $\alpha$ -iodomalnonitrile,<sup>1i,j</sup> it was not applicable to the reaction with electron-rich alkenes

because of the nature as an electrophilic iodine source of unstable allyl- $\alpha$ -iodoactive methines (Eq. (2)).<sup>2</sup> In this paper, we report radical iodine atom transfer [3+2]-cycloaddition with various alkenes using dimethyl 2-(iodomethyl)cyclopropane-1,1-dicarboxylate **1** as a new homoallyl radical precursor (Eq. (3)).

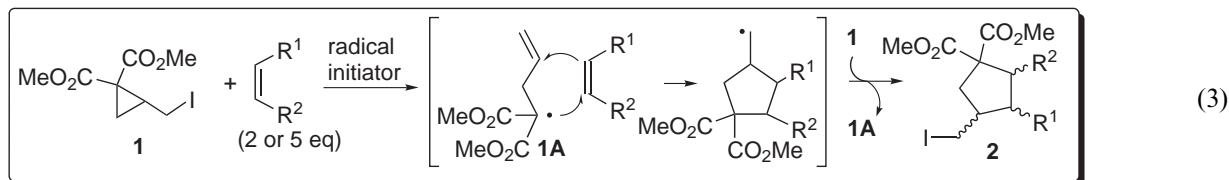
We expected that dimethyl 2-(iodomethyl)cyclopropane-1,1-dicarboxylate **1** may efficiently produce an allylmalonate radical **1A** via regioselective cleavage of the C–C(CO<sub>2</sub>Me)<sub>2</sub> bond of a cyclopropylmethyl radical, and the resulting **1A** would possibly give iodoalkylated cyclopentane derivatives through radical iodine atom transfer reaction with alkenes (Eq. (3)).<sup>3</sup> In addition, the reagent **1**, which does not have an alkenyl group and does not work as an electrophilic iodine source, may be an efficient homoallyl radical precursor. That is, the use of radical precursor **1** not having an alkenyl group such as that in vinylcyclopropane (Eq. (1))<sup>1g</sup> and allyl- $\alpha$ -iodoactive methines (Eq. (2))<sup>1h-j</sup> may bring about an increase in the chemical yield, because the attack of the resulting homoallyl radical to the precursor should be ignored. Furthermore, since this stable (iodomethyl)cyclopropane **1** does not have the nature as an electrophilic iodine source such as unstable allyl- $\alpha$ -iodoactive methines (Eq. (2)), the reaction with electron-rich alkene may be also possible.<sup>2</sup> (Iodomethyl)cyclopropane **1** can be easily prepared through iodocarbocyclization reaction of allylmalonate which was previously found by our group.<sup>4</sup>



**Keywords:** radicals and radical reactions; cycloadditions; cyclopropanes; cyclopentanes.

\* Corresponding author. Tel./fax: 81-426-76-3257; e-mail: taguchi@ps.toyaku.ac.jp

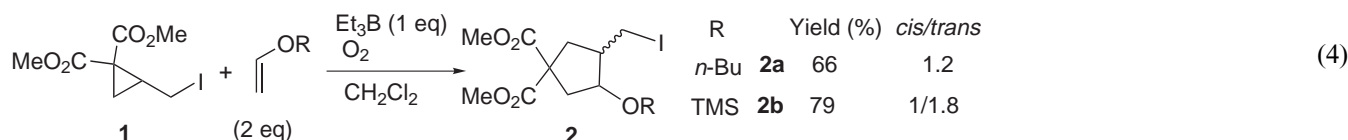
Radical iodine atom transfer [3+2]-cycloadditions of **1** with various alkenes such as electron-rich alkene, 1-alkene and 1,1-, 1,2-disubstituted alkene were examined



(Eq. (4) and Table 1). The reaction of **1** with electron-rich alkene (2 equiv.) such as alkyl vinyl ether and silyl enol ether smoothly proceeded in the presence of Et<sub>3</sub>B (1 equiv., O<sub>2</sub> atmosphere) to give functionalized cyclopentane derivatives **2a** and **2b** in good yields (66, 79%), respectively, while the diastereoselectivities of these reactions were low (Eq. (4)).<sup>5</sup>

In contrast to the reaction with enol ether derivatives,

reaction of **1** with 1-hexene gave the product **2c** in poor yield (22%) under the same conditions. On the other hand, the addition of Yb(OTf)<sub>3</sub> was found to lead to the improvement of the chemical yield.<sup>6,7</sup> That is, the reaction of **1** with 1-hexene (2 equiv.) in the presence of Et<sub>3</sub>B (1 equiv.) and Yb(OTf)<sub>3</sub> (1 equiv.) under an O<sub>2</sub> atmosphere gave **2c** in good yield (82%) with high diastereoselectivity (*cis/trans* = 11.2) (Table 1, Entry 1).<sup>8</sup> Yb(OTf)<sub>3</sub> may promote both the rate of addition of the



**Table 1.** Radical [3+2]-cycloaddition with (iodomethyl)cyclopropane **1**

<b>1</b> + Alkene (2 eq)		Et <sub>3</sub> B (1 eq), Yb(OTf) <sub>3</sub> (1eq), O <sub>2</sub> CH <sub>2</sub> Cl <sub>2</sub> → Products <b>2</b>			
Entry	Alkene	Products	Yield (%) <sup>a</sup>	<i>cis/trans</i>	
1			<b>2c</b> 82	11.2 <sup>b</sup>	
2			<b>2d</b> 67	6.0 <sup>b</sup>	
3			<b>2e</b> 68	10.8 <sup>b</sup>	
4			<b>2f</b> 88	—	
5			<b>2g</b> 73	—	
6 <sup>c</sup>			<b>2h</b> 70	> 50	
7 <sup>c</sup>			<b>2i</b> 65	2.2 <sup>d,e</sup>	

<sup>a</sup> Isolated yield.

<sup>b</sup> The ratio was determined by <sup>13</sup>C-NMR.

<sup>c</sup> 5 eq of alkene was used.

<sup>d</sup> Two stereoisomers were mainly obtained, while the stereochemistries were not determined.

<sup>e</sup> The ratio was determined by <sup>1</sup>H-NMR.

malonate radical and the iodine abstraction through the formation of bidentate complex with the malonate.<sup>7b,c,9</sup> The reaction with other 1-alkenes such as vinylsilane and homoallyl alcohol also proceeded with *cis*-selectivity (**2d**: *cis/trans*=6, **2e**: *cis/trans*=10.8) to give product **2d** and **2e** in good yields (67 and 68%, Entries 2,3). In the reactions with 1-hexene and vinylsilane, considerable increase in both the chemical yields and diastereoselectivities was observed in comparison with those attained by Curran's method (**2c**: 53%, *cis/trans*=4, **2d**: 46%, *cis/trans*=2) which gives the same products **2c** and **2d** on using allyliodomalonate.<sup>1h</sup> Similar to that with 1-alkenes, in the cycloaddition reaction with 1,1-disubstituted alkenes, products **2f** and **2g** were obtained in good yields (88 and 73%), respectively (Entries 4,5).

The present reaction is applicable not only to 1-alkenes and 1,1-disubstituted alkenes but also to less reactive 1,2-disubstituted alkenes; that is, the reaction with cyclopentene and 3-hexene also gave the products **2h**<sup>5</sup> and **i** in good yields (70 and 65%), respectively (Entries 6,7). These results should be noteworthy, because it has been reported that the reaction with 1,2-disubstituted alkene hardly proceeds under Curran's conditions using allyliodomalonate.<sup>1h</sup> Thus, we found radical iodine atom transfer [3+2] cycloaddition of homoallyl radical species which can be applied to a wide range of 1-alkene, 1,1-, 1,2-disubstituted alkenes and electron-rich alkenes.

In conclusion, we have succeeded in the development of radical iodine atom transfer [3+2] cycloaddition with various alkenes using dimethyl 2-(iodomethyl)cyclopropane-1,1-dicarboxylate as a new homoallyl radical precursor. The present reaction should provide an efficient and practical synthetic method of functionalized cyclopentane skeletons.

## References

1. The reaction with electron-deficient alkenes: (a) Angoh, A. G.; Clive, D. L. J. *J. Chem. Soc., Chem. Commun.* **1985**, 980; (b) Cekovik, Z.; Saicic, R. *Tetrahedron Lett.* **1986**, 27, 5893; (c) Curran, D. P.; Chen, M. *J. Am. Chem. Soc.* **1987**, 109, 6558; (d) Barton, D. H. R.; Zard, S. Z.; da Silva, E. *J. Chem. Soc., Chem. Commun.* **1988**, 285; (e) Feldman, K. S.; Romanelli, A. L.; Ruckle, R. E. Jr.; Miller, R. F. *J. Am. Chem. Soc.* **1988**, 110, 3300; (f) Curran, D. P. In *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I., Eds.; Pergamon Press: New York, 1991; Vol. 4, pp. 779 and references cited therein. The reaction with simple alkenes: (g) Miura, K.; Fugami, K.; Oshima, K.; Utimoto, K. *Tetrahedron Lett.* **1988**, 29, 5135; (h) Curran, D. P.; Chen, M.; Spletzer, E.; Seong, C. M.; Chang, C. J. *J. Am. Chem. Soc.* **1989**, 111, 8872; (i) Curran, D. P.; Seong, C. M. *Tetrahedron* **1992**, 48, 2157; (j) *Idem. ibid.* **1992**, 48, 2175.
2. Curran, D. P.; Eichenberger, E.; Collis, M.; Roepel, M. G.; Thoma, G. *J. Am. Chem. Soc.* **1994**, 116, 4279.
3. For reviews in relation to the reaction of carbon centered radicals: (a) Giese, B. *Radicals in Organic Synthesis: Formation of Carbon–Carbon Bonds*; Pergamon Press: New York, 1986; (b) Curran, D. P. *Synthesis* **1988**, 417; (c) Curran, D. P. *Synthesis* **1988**, 489; (d) Jasperse, C. P.; Curran, D. P.; Fevig, T. L. *Chem. Rev.* **1991**, 91, 1237; (e) Motherwell, W. B.; Crich, D. *Free Radical Chain Reactions in Organic Synthesis*; Academic Press: London 1992; (f) Fossey, J.; Lefort, D.; Sorba, J. *Free Radicals in Organic Chemistry*; John Wiley & Sons, Inc.: New York, 1995; (g) Curran, D. P.; Porter, N. A.; Giese, B. *Stereochemistry of Radical Reactions*; VCH: New York, 1996.
4. Inoue, T.; Kitagawa, O.; Ochiai, O.; Taguchi, T. *Tetrahedron: Asymmetry* **1995**, 6, 691.
5. The stereochemistries of the products were determined on the basis of NOESY measurements.
6. Typical reports in relation to Lewis acid-mediated radical reaction: (a) Guindon, Y.; Yoakim, C.; Lemieux, R.; Boisvert, L.; Delorme, D.; Lavallee, J. F. *Tetrahedron Lett.* **1990**, 31, 2845; (b) Yamamoto, Y.; Onuki, S.; Yamamoto, M.; Asao, N. *J. Am. Chem. Soc.* **1994**, 116, 421; (c) Renaud, P.; Moufid, N.; Kuo, L. H.; Curran, D. P. *J. Org. Chem.* **1994**, 59, 3259; (d) Nishida, M.; Ueyama, E.; Hayashi, H.; Ohtake, Y.; Yamamura, Y.; Yanaginuma, E.; Yonemitsu, O.; Nishida, A.; Kawahara, N.; *J. Am. Chem. Soc.* **1994**, 116, 6455; (e) Urabe, H.; Yamashita, K.; Suzuki, K.; Kobayashi, K.; Sato, F. *J. Org. Chem.* **1995**, 60, 3576; (f) Sibi, M. P.; Ji, J.; Wu, J. H.; Gürtler, S.; Porter, N. A.; *J. Am. Chem. Soc.* **1996**, 118, 2000; (g) Murakata, M.; Jono, T.; Mizuno, Y.; Hoshino, O. *J. Am. Chem. Soc.* **1997**, 119, 11713. For reviews: (h) Renaud, P.; Gerster, M. *Angew. Chem., Int. Ed. Engl.* **1998**, 37, 2562; (i) Guindon, Y.; Jung, G.; Guerin, B.; Ogilvie, W. W. *Synlett* **1998**, 213.
7. Among various Lewis acids examined [TiCl<sub>4</sub>, SnCl<sub>4</sub>, ZnCl<sub>2</sub>, MgBr<sub>2</sub>, Sc(OTf)<sub>3</sub>, Yb(OTf)<sub>3</sub>], the use of Yb(OTf)<sub>3</sub> gave the best result. Reports in relation to Yb(OTf)<sub>3</sub>-promoted radical reaction: (a) Sibi, M. P.; Jasperse, C. P.; Ji, J. *J. Am. Chem. Soc.* **1995**, 117, 10779; (b) Mero, C. L.; Porter, N. A. *J. Am. Chem. Soc.* **1999**, 121, 5155; (c) Yang, D.; Ye, X.; Gu, S.; Xu, M. *J. Am. Chem. Soc.* **1999**, 121, 5579.
8. General procedure of radical [3+2] cycloaddition reaction of dimethyl 2-(iodomethyl)cyclopropane-1,1-dicarboxylate **1** with alkenes. Under an argon atmosphere, to a solution of dimethyl 2-(iodomethyl)cyclopropane-1,1-dicarboxylate **1** (149 mg, 0.5 mmol), 1-hexene (0.13 mL, 1 mmol) and Yb(OTf)<sub>3</sub> (310 mg, 0.5 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (4 mL) was added Et<sub>3</sub>B (0.5 mL, 1 M hexane solution). After being stirred for 5 h at 0°C under an O<sub>2</sub> atmosphere, the mixture was poured into aqueous NH<sub>4</sub>Cl solution and extracted with Et<sub>2</sub>O. The Et<sub>2</sub>O extracts were washed with brine, dried over MgSO<sub>4</sub>, and evaporated to dryness. Purification of the residue by silica-gel column chromatography (hexane/AcOEt=20) gave an inseparable mixture of *cis*-**2c** and *trans*-**2c** (157 mg, 82%, *cis/trans*=11.2).
9. In the reaction with alkyl vinyl ether and silyl enol ether, addition of Yb(OTf)<sub>3</sub> resulted in the formation of a complex mixture.